Iron (III) Chloride, Anhydrous



Section 1

Product Description

Product Name: Iron (III) Chloride, Anhydrous **Recommended Use:** Science education applications

Synonyms: Ferric Chloride

Distributor: Carolina Biological Supply Company

2700 York Road, Burlington, NC 27215

1-800-227-1150

Chemical Information: 800-227-1150 (8am-5pm (ET) M-F)

Chemtrec: 800-424-9300 (Transportation Spill Response 24 hours)

Section 2

Hazard Identification

Classification of the chemical in accordance with paragraph (d) of §1910.1200;

DANGER







May be corrosive to metals. Harmful if swallowed. Causes skin irritation. Causes serious eye damage. Toxic to aquatic life. Toxic to aquatic life with long lasting effects.

GHS Classification:

Substance or mixture corrosive to metals Category 1, Serious Eye Damage/Eye Irritation Category 1, Skin Corrosion/Irritation Category 2, Hazardous to the aquatic environment - Acute Category 2, Hazardous to the aquatic environment - Chronic Category 2, Acute Toxicity - Oral Category 4

Acute Toxicity Dermal Contains
Acute Toxicity Inhalation Gas

100 % of the mixture consists of ingredient(s) of unknown toxicity
100 % of the mixture consists of ingredient(s) of unknown toxicity

Contains

Acute Toxicity Inhalation Vapor 100 % of the mixture consists of ingredient(s) of unknown toxicity

Contains

Acute Toxicity Inhalation Dust/Mist 100 % of the mixture consists of ingredient(s) of unknown toxicity

Contains

Section 3

Composition / Information on Ingredients

Chemical NameCAS #%Iron (III) Chloride, Anhydrous7705-08-0100

Section 4

First Aid Measures

Emergency and First Aid Procedures

Inhalation: In case of accident by inhalation: remove casualty to fresh air and keep at rest.

Eyes: IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy

to do. Continue rinsing.

Skin Contact: IF ON SKIN: Wash with plenty of soap and water. If skin irritation occurs: Get medical advice/attention.

Take off contaminated clothing and wash before reuse.

Ingestion: IF SWALLOWED: Call a POISON CENTER or doctor/physician if you feel unwell.

Section 5

Firefighting Procedures

Extinguishing Media: Use dry chemical, CO2 or appropriate foam.

Fire Fighting Methods and Protection: Firefighters should wear full protective equipment and NIOSH approved self-contained

breathing apparatus.

Fire and/or Explosion Hazards: Fire or excessive heat may produce hazardous decomposition products. Liberates

hydrochloric acid fumes when damp.

Hazardous Combustion Products: Hydrogen chloride

Section 6

Spill or Leak Procedures

Steps to Take in Case Material Is Released or Spilled:

Exposure to the spilled material may be irritating or harmful. Follow personal protective equipment recommendations found in Section 8 of this SDS. Additional precautions may be necessary based on special circumstances created by the spill including; the material spilled, the quantity of the spill, the area in which the spill occurred. Also consider the expertise of employees in the area responding to the spill.

Prevent the spread of any spill to minimize harm to human health and the environment if safe to do so. Wear complete and proper personal protective equipment following the recommendation of Section 8 at a minimum. Dike with suitable absorbent material like granulated clay. Gather and store in a sealed container pending a waste disposal evaluation.

Absorb spillage to prevent material damage.

Collect spillage.

Section 7

Handling and Storage

Handling: Keep only in original container. Wash thoroughly after handling. Do no eat, drink or smoke when using this

product. Avoid release to the environment. Wear protective gloves/protective clothing/eye protection/face protection. Keep container tightly closed in a cool, well-ventilated place. Readily absorbs moisture from air. Avoid direct sunlight and heat. Keep away from water and ice Avoid contact with skin and eyes. Harmful by

inhalation and if swallowed.

Storage: Store in corrosive resistant/... container with a resistant inner liner. Keep container tightly closed and dry.

Storage Code: White - Corrosive. Separate acids from bases; separate oxidizer acids from organic acids.

Section 8

Protection Information

 Chemical Name
 (TWA)
 (STEL)
 (TWA)
 (STEL)

 Iron (III) Chloride, Anhydrous
 1 mg/m3 TWA (as Fe)
 N/A
 N/A
 N/A

Control Parameters

Respirator Type(s): Eye Protection:

Engineering Measures: Local exhaust ventilation or other engineering controls are normally required when

handling or using this product to avoid overexposure.

Personal Protective Equipment (PPE): Lab coat, apron, eye wash, safety shower.

Respiratory Protection:

Respiratory protection may be required to avoid overexposure when handling this

product. General or local exhaust ventilation is the preferred means of protection. Use a respirator if general room ventilation is not available or sufficient to eliminate symptoms. NIOSH approved air purifying respirator with acid gas cartridge and dust/mist filter Wear chemical splash goggles when handling this product. Have an eye wash station

available.

Skin Protection: Avoid skin contact by wearing chemically resistant gloves, an apron and other protective

equipment depending upon conditions of use. Inspect gloves for chemical break-through and replace at regular intervals. Clean protective equipment regularly. Wash hands and other exposed areas with mild soap and water before eating, drinking, and when leaving

work.

Gloves: Butyl rubber, Neoprene, Nitrile

Section 9

Physical Data

Formula: FeCl3

Molecular Weight: 162.22

Appearance: Red-brown Crystalline Solid

Odor: No data available

Odor Threshold: No data available

pH: No data available **Melting Point:** 304 C

Boiling Point: No data available Flash Point: No data available

Flammable I imits in Air: No data available

Vapor Pressure: 1.33 hPa at 194 °C

Evaporation Rate (BuAc=1): 1.33 hPa at 194 °C (solid)

Vapor Density (Air=1): 5.6 Specific Gravity: 2.90 Solubility in Water: Soluble

Log Pow (calculated): No data available
Autoignition Temperature: No data available
Decomposition Temperature: No data available

Viscosity: No data available

Percent Volatile by Volume: No data available

Section 10 Reactivity Data

Reactivity: Not generally reactive under normal conditions.

Chemical Stability: Stable under normal conditions.

Conditions to Avoid: Exposure to moisture

Incompatible Materials: Contact with (specify material) may form shock-sensitive materials, Potassium Metal,

Sodium Metal

Hazardous Decomposition Products: Hydrogen chloride Hazardous Polymerization: Hydrogen chloride Will not occur

Section 11 Toxicity Data

Routes of Entry Inhalation and ingestion.

Symptoms (Acute): Respiratory disorders, Increased Respiration, Tachycardia, Hypoxemia (low blood oxygen), Metabolic

Acidosis, Vomiting, Heart attack

Delayed Effects: No data available

Acute Toxicity:

Chemical NameCAS NumberOral LD50Dermal LD50Inhalation LC50Iron (III) Chloride, Anhydrous7705-08-0Oral LD50 MouseNot determinedNot determined

200 mg/kg

Oral LD50 Rat 316

mg/kg

Carcinogenicity:

Chemical NameCAS NumberIARCNTPOSHAIron (III) Chloride, Anhydrous7705-08-0Not listedNot listedNot listed

Chronic Effects:

Mutagenicity: No evidence of a mutagenic effect.

Teratogenicity: No evidence of a teratogenic effect (birth defect).

Sensitization: No evidence of a sensitization effect.

Reproductive: Evidence of negative reproductive effects in males.

Target Organ Effects:

Acute: Cardiovascular system
Chronic: No information available

Section 12 Ecological Data

Overview: Severe ecological hazard. This product may be toxic to plants and/or wildlife.

Mobility: This material is expected to have only slight mobility in soil. It absorbs strongly to most soil types.

Persistence: Adsorbs to soil.

Bioaccumulation: Bioconcentration may occur.

Degradability: No data
Other Adverse Effects: No data

Chemical Name CAS Number Eco Toxicity

Iron (III) Chloride, Anhydrous 7705-08-0 96 HR LC50 GAMBUSIA AFFINIS 75.6 MG/L [STATIC]

48 HR EC50 DAPHNIA MAGNA 27.9 MG/L

48 HR EC50 DAPHNIA MAGNA 9.6 MG/L [STATIC]

Section 13 Disposal Information

Disposal Methods: Dispose in accordance with all applicable Federal, State and Local regulations. Always

contact a permitted waste disposer (TSD) to assure compliance.

Waste Disposal Code(s): Not Determined

Section 14 Transport Information

Ground - DOT Proper Shipping Name:

UN1773

Ferric Chloride, Anhydrous

Class 8 P.G. III

Air - IATA Proper Shipping Name:

UN1773

Ferric Chloride, Anhydrous

Class 8 P.G. III

Section 15 **Regulatory Information**

TSCA Status: All components in this product are on the TSCA Inventory.

Chemical Name CAS § 313 Name **CERCLA RQ CAA 112(2)** § 304 RQ § 302 TPQ

Number

Iron (III) Chloride, Anhydrous 7705-08-0 No 1000 lb 1000 lb final No No

> RQ RQ; (454 kg)

Section 16 Additional Information

Printed: 10-29-2015 Revised: 09/09/2015 Replaces: 09/03/2014

The information provided in this (Material) Safety Data Sheet represents a compilation of data drawn directly from various sources available to us. Carolina Biological Supply makes no representation or guarantee as to the suitability of this information to a particular application of the substance covered in the (Material) Safety Data Sheet.

Glossary

ACGIH American Conference of Governmental NTP National Toxicology Program Occupational Safety and Health Administration

Industrial Hygienists **OSHA**

CAS Chemical Abstract Service Number PEL Permissible Exposure Limit

Comprehensive Environmental Response, **CERCLA** Parts per million ppm

Compensation, and Liability Act **RCRA**

Resource Conservation and Recovery Act U.S. Department of Transportation

DOT SARA Superfund Amendments and Reauthorization Act

IARC International Agency for Research on Cancer TLV Threshold Limit Value

N/A Not Available **TSCA** Toxic Substances Control Act

IDLH Immediately dangerous to life and health