## Fisher Science Education

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Effective Date: MSDS No.: January 1, 2007 SS0290

24 HOUR EMERGENCY ASSISTANCE

SECTION	NAME		24 HOU
Product	Sodium Bisulfate		>
Chemical Synonyms	Sodium Acid Sulfate, Sodium Hydrogen Sulfate	Sulfate	$\bigcirc$
Formula	NaHSO <sub>4</sub>		 [<
Unit Size	up to 2.5 Kg.		HAZARI
C.A.S. No.	7681-38-1		OMAL

Day 585-226-6177 CHEMTREC Fire Health Reactivity 0 0 ω

SECTION II INGREDIENTS OF MIXTURES AD RATING MODERATE SERIOUS 3

SIMH

SEVERE

Principal Component(s) DANGER! CORROSIVE! CAUSES SEVERE BURNS TO SKIN, EYES AND Sodium bisulfate 100% % TLV Units See Section V.

MUCOUS MEMBRANES. HARMFUL IF SWALLOWED OR INHARED. MOISTURE SENSITIVE.

SECTION III	PHYSICAL DATA		
Melting Point (°F)	177°C (350°F)	Specific Gravity (H <sub>2</sub> O = 1) 2.435 @ 13°C	2.435 @ 13°C
Boiling Point (°F)	Decomposes.	Percent Volatile by Volume (%)	N/A
Vapor Pressure (mm Hg) N/A	N/A	Evaporation Rate (n-Butyl acetate =1)	A/N
Vapor Density (Air=1)	N/A		
Solubility in Water	Soluble.		

Extinguisher Flash Point Method Used) SECTION IV Non-combustible White crystalline powder, globular, crystals; no odor. Hygroscopic **FIRE AND EXPLOSION HAZARD** Flammable Limits in Air DATA

Upper

Use any media suitable for extinguishing supporting fire

Appearance & Odor

PROCEDURES SPECIAL FIREFIGHTING

apparatus and full protective clothing to prevent contact with skin In fire conditions, wear a NIOSH/MSHA self-contained breathing and eyes.

(1996 EMERGENCY RESPONSE GUIDEBOOK, DOT P 5800.7, GUIDE PAGE NO. 154

EXPLOSION HAZARDS UNUSUAL FIRE AND

toxic furnes of oxides of sulfur; will react with water or steam to solution. Dangerous; when heated to decomposition, it emits highly Product readily dissolves in water to form weak Sulfuric acid Sulfuric acid is highly corrosive and causes severe burns on contact produce hear and toxic furnes

D.O.T. Approved by U.S. Department of Labor "essentially similar" to form OSHA-20 Corrosive solids, n.o.s., (Sodium hydrogen sulfate), 8, UN1759, PG III

HEALTH HAZARD DATA

Threshold Limited Value

None established. Estimated human oral lethal dose is 10 to 30 grams Oral LD<sub>50</sub> Rat: 2800 mg/kg; Eye irritation Rabbit - severe irritation

Effects of Overexposure

INGESTION: May irritate or burn mouth, esophagus or stomach. EYES: throat or lungs. Target organs: None known. Dust may cause upper respiratory tract irritation. May irritate or burn nose Contact with powder may cause severe burns. SKIN: Prolonged or repeated skin contact may cause skin irritation and/or burns. INHALATION

**Emergency and** First Aid Procedures

vomiting only if advised by appropriate medical personnel. Never give INGESTION: Call physidian or Poison Control Center immediately. Induce

medical attention. INHALATION: Remove to fresh air. If not breathing, give artificial respiration. If breathing is anything by mouth to an unconscious person. **EYES**: Check for and remove contact lenses. Flush thoroughly SKIN: Remove contaminated clothing. Flush thoroughly with mild soap and water. If irritation occurs, get with water for at least 15 minutes, lifting upper and lower eyelids occasionally. Get immediate medical attention difficult, give oxygen. Get medical attention.

SECTION VI Stability Unst	N VI Unstable Stable	×	REACTIVITY DATA Conditions to Avoid	0
- Carriery	Stable	×		Excessive temperature and heat
Incompatibility (Materials to Avoid)	bility to Avoid)	Alkalie Sulfuri	Alkalies and strong oxidizers a Sulfuric acid solution.	Alkalies and strong oxidizers and permanganates. Reacts with water to form weak Sulfuric acid solution.
Hazardous Decomposi	Hazardous  Decomposition Products	lucts	When heated to do	When heated to decomposition emits toxic fumes of oxides of sulfur
Hazardous	Hazardous Polymerization	ion	Conditions to Avoid	oid
May Occur		Will Not Occur		Not applicable.
	,	`		-

Waste Disposal Method

Steps to be taken in case

SECTION VII

OR LEAK PROCEDURES

material is released or spilled

Discharge, treatment, or disposal may be subject to Federal, State or Local laws. These disposal guidelines are intended for the disposal of catalog-size quantities only. copious amounts of water

sweep up and place in a suitable container for proper disposal Wearing proper safety equipment and using adequate ventilation, Dissolve in water, neutralize with soda ash and flush to sewer with

If federal, state and local regulations permit, dissolve in water, neutralize with

soda ash and flush to sewer with copious amounts of water.

SECTION VIII SPECIAL PROTECTION INFORMATION

Protective Gloves Ventilation espiration Protection Local Exhaust None needed in normal laboratory handling. hood or wear a NIOSH/MSHA approved dust mask or respirator Rubber. Recommended Recommended. Eye Protection Special conditions prevail, work in ventilation Chemical safety goggles. 6

Other Protective

SECTION IX Goggles, smock, apron, eye wash station, proper gloves, ventilation hood SPECIAL PRECAUTIONS

keep container tightly closed when not in use Precautions to be Taken Handling & Storing

Wash thoroughly after handling Store in a cool, dry place.

Other Precautions Head label on container before using. Do not wear contact lenses when working with chemicals For laboratory use only. Not for drug, food or household use. Keep out of reach of children.

ω Date 01/01/07 Hygroscopic. Avoid contact with eyes, skin and clothing. Use with adequate ventilation. Remove and wash contaminated clothing Chemical Safety Coordinator

The mindration contained freem is furnished without warranty of any find. Employees Should use this information only as a supplement to other information pathered by them and must make interpreted interpreted in completeness of information from all sources to assure proper use of these materials and the safety and health of employees. \* Hazarridous Materials industrial Standards. Printed on recycled paper.

Approved

James A. Bertsch

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Revision No.